Observation and Measurement of the Environment

				_		
T	OBSERVATION -	_	interaction	of	OUT	senses
_	ADARIZATION.		111111111111111111111111111111111111111			

with our environment

- A. The five senses include: (1) 5 ight



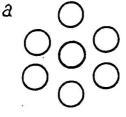


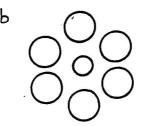


- (3) touch / Reel
- (4) taste
- (5) <u>smell</u>

B. Testing your powers of observation:

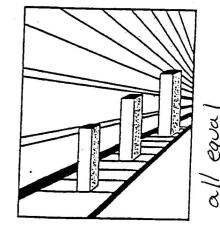
- 1. Use only your sense of sight to make observations to determine:
- (a.) Which, if any, of the three center circles is largest or smallest (a, b or c)?
- (b.) Which if any of the three blocks is the tallest (front, middle, or back)?



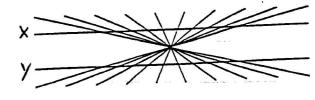


Largest





(C.) Which if any of the labeled pairs of lines are parallel (ab or xy)?



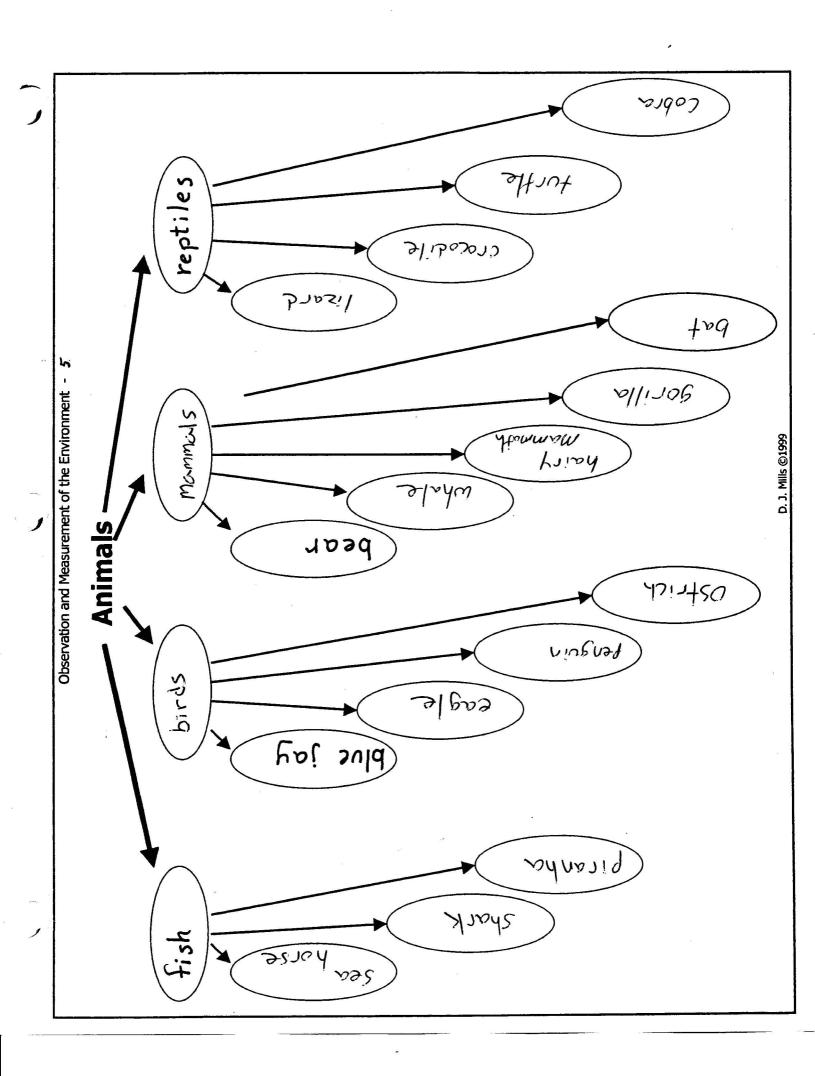
both parallel

2. How can we determine if our power of observation using only sight was accurate? <u>Measure with ruler</u>											
3. Check to determine if you were accurate. How many observations using only sight did you have correct(1,2,or 3)?											
C. Our powers of observation are limited by our senses.											
our powers of observation. These devices have been invented by people to extend the human senses beyond their normal limits, and thus enable us to make observations that would otherwise be impossible or highly inaccurate. E. Examples: 4. Wind vane 3. Spring Scale 3. Spring Scale 4. Wind vane											
1. graduated 7. telescope cylinder 7. telescope											
5. Compass 6. triple hearn balance 10. ruler											
9. Magnifying glass											
F. Which instruments can be used to accurately measure each of											
these?											
1. microscope 2. telescope 3. thermometer triple beam balance											

TRII	FERENCE - an interpretation of an observation
TIAI	-ENERGE - an interpretation of an observation
_	based on one's knowledge
_	and or experience,
Α.	For example, if you observed the appearance of more and more cloud
	that were darker and darker in color, you might infer that
_	if will rain
В.	List the observations and inferences that you can make based on this
	picture:
	Fy / John
	1. observations: 4 dogs, I large and 3 small
	puppies feedings
	2. inferences: Large dog is mother
	puppies are hungry
_	The life and attachment as either an observation or an inference:
C.	Identify each statement as either an observation or an inference:
	1. (a.) The dog is growling.
	(b.) The dog is angry
	(b.) The pebble is smooth and rounded
	(c.) The pebble was carried by a saled in (c.) The pebble is light brown in color
	3. (a.) By tomorrow, the stream will overflow its banks
	(b.) The river is high, muddy and flowing swiftly.
	(c.) The rainfall has been continuous and is very heavy.
	4. (a.) The tire has a leak
	(b.) The tire is flat.
	5. (a.) There is a track on this trail.
	(b.) The track was made by a large deer
	6. (a.) The leaves on the trees are moving.
	(b.) The leaves on the trees are moving, so the wind must be blowing

D. J. Mills @1999

SHARK



IV. MEASUREMENT

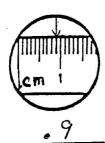
- A. A measurement is a way of expressing an observation with greater precision. It provides a numerical value for some property of the object or the event being observed.
 - 1. All measurements consist of: (a.) <u>Numerical value</u>
 (b.) <u>label</u>
 - 2. Properties that can be measured include length, area, volume, mass, weight, temperature, density, time, etc.
- B. <u>Linear Measurement</u> (or "one dimensional measurement") ______
 - 1. Instrument ruler
 - 2. Unit meter
 - (a.) centi 1/100 meter 1/00 centimeters = 1 meter
 - (b.) milli 1/1000 meter /000 millimeters = 1 meter
 - (c.) kilo 1000 meters / kilometer = 1000 meters
 - 3. Using the model below, give the value of each dot to the nearest tenth of a centimeter, and then convert to millimeters.

NOT TO SCALE

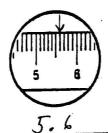
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111111	1	1		ببادي		[""	
cm	İ		2		3	2	

- (a.) A 2.5 centimeters, or 25 millimeters
- (b.) B $\underline{}$ centimeters, or $\underline{}$ millimeters
- (c.) C <u>4.0</u> centimeters, or <u>40</u> millimeters
- (d.) D / / centimeters, or // millimeters
- (e.) E / 6 centimeters, or /6 millimeters
- (f.) F = 3.4 centimeters, or = 34 millimeters
- (g.) G 2.0 centimeters, or 20 millimeters
- 4. What is the length indicated by the arrow in each of the following pictures?

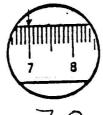
(a.)



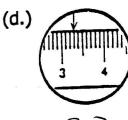
(b.)



(c.)



7.0

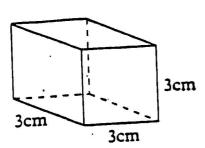


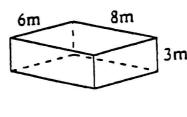
3.3

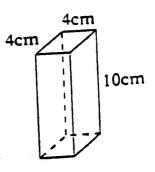
	·
5. Measure each line segment to t	ion and Measurement of the Environment - 7 he nearest tenth of a centimeter.
and then convert to millimet	
4, 1 (a.)	$\frac{40-42}{2}$ cm, or $\frac{40-42}{2}$ mm
5.6 (b.).	5,5-5,1 cm, or 55-57 mm
3.9 (c.)	<u>3.8 - 40</u> cm, or <u>38 − 40</u> mm
6.4 (d.)	6.3 - 6.5 cm, or <u>63-65</u> mm
/, ○ (e.) ——	$\frac{9-1.1}{100}$ cm, or $\frac{9-11}{100}$ mm
C. Area (or "two-dimensional measurement	ent") - amount of
Surface space	
1. Instrument - <u>ruler</u>	•
2. Formula: $A = 1 \times \omega$	
3. Units – "square" units	
(a.) sq. cm or <u></u> <u> </u>	
(b.) sq. m or	
(c.) sq. km or <u>km²</u>	
4. Determine the area of each figu	re below.(They have been drawn scale.)
(a.)	(b)
8cm ²	
0 (71)	
	20cm2
2 14 MAIN 2000 C 150	
	D 1 Mills @1999

Observation and Measurement of the Environment - 8 D. <u>Volume</u> (or "three-dimensional measurement" sometimes referred to
as the "size" of an object) - amount of
space an object occupies
1. Volume of Liquids
(a.) Instrument - graduated cylinder
(b.) Units - milliliters
(C.) 100
Measurement: 50
(d.) What is the volume of the liquid in the graduated cylinders illustrated below?
2. Volume of Rectangular Solids
(a.) Instrument:
(b.) Formula: $Y = / \times \omega \times h$
(c.) Units – "cubic" units
(1.) cu. cm or \underline{CC} or \underline{Cm}^3
(2.) cu. m or or
(d.) cubic centimeter = milliliter

(e.)Determine the volume of the objects in each illustration:







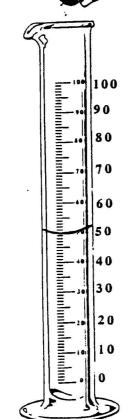
27 cm3

144 cm 3

160 cm3

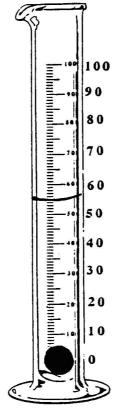
3. Volume by Displacement





Volume of Water

50ml



Volume of Water and Marble

55ml

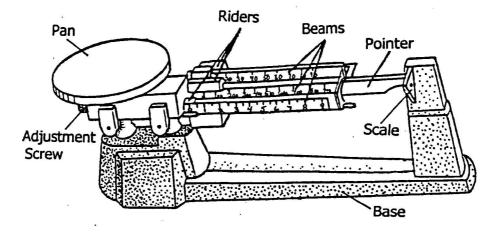
Volume of Marble

5m/

E. Mass - the amount of matter an

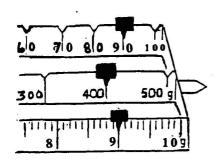
object possesses

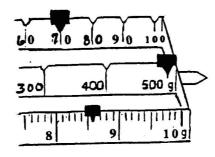
1. Instrument - triple beam balance

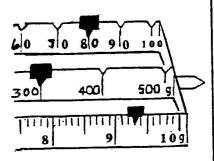


2. Units: (a.) <u>Grams</u>
(b.) <u>Kilograms</u>

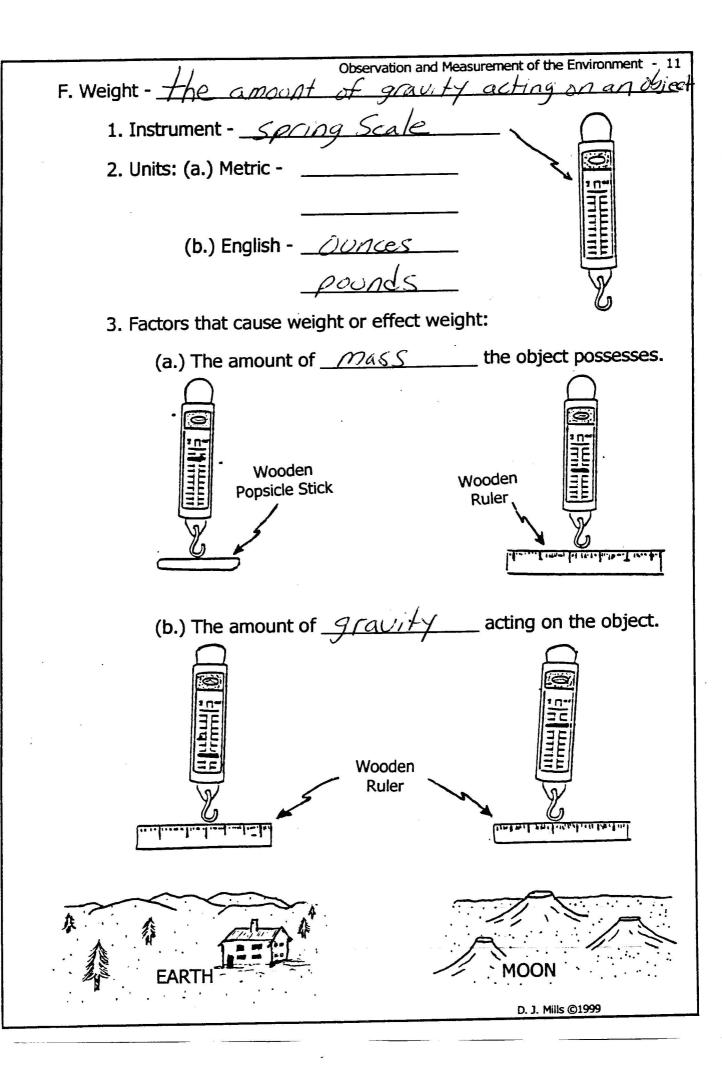
3. Determine the mass by reading the scales of the beams in the illustrations below:





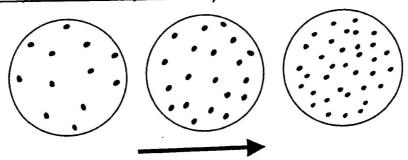


(a.) 499.09 (b.) 578.69 (c.) 389.39



G. Density - Concentration of matter in an object;

ratio of mass per unit volume



1. Formula:

$$D = \frac{M}{V}$$

page _/ the Earth Science Reference Tables

- 2. Instruments:
 - (a.) Density of a liquid

graduated cylinder

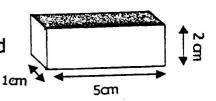
(b.) Density of a solid

- ruler

(b.) g/cm^3

4. Solve the density problems below:

(a.) The dimensions of a rectangular solid object is given as illustrated to the right. Given that this



object has a mass of 150 grams,

determine the density of the object. $\frac{15 \text{ g/cm}^3}{2}$

(b.) In the illustration to the right,

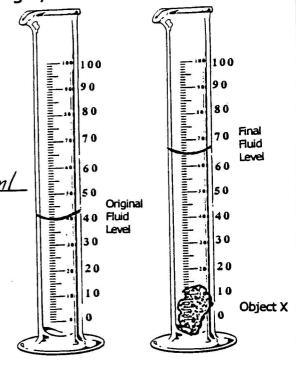
the mass of object X is 80 grams. The volume can be determined by the change in the fluid level.

What is the density of Object X? 3.2 9/ml

$$D = \frac{m}{25ml}$$

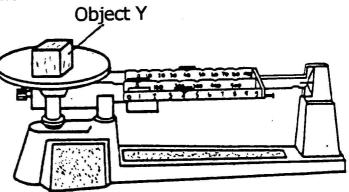
$$D = \frac{80}{25ml}$$

D= 3.2 9/m1



(c.) Object Y is a perfect cube. The density of object Y is 5.5 g/cm3

D= m/V 5.5 = 44

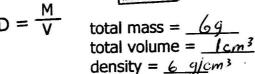


- (1.) What is the mass of object Y as shown by the position of the balance riders? ____44
- (2.) Calculate the volume of object Y. $8cm^3$
- (3.) Since object Y is a perfect cube, determine the length of each side of that cube. 2 cm

H. Density Relationships

- 1. Density is Relation to Size
 - (a.) Model Problem



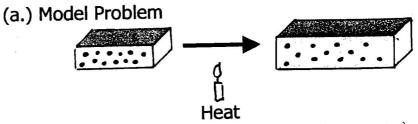


total mass = 24gtotal volume = $4cm^3$ density = $6g/cm^3$

(b.)	Relations						
	does	not	det	ermin	e/at	ffect	
	175	der	osit	Y			
(c.)	. t	↑					
	Densit					¥1	
	۵						
			<u> </u>				
			Size				

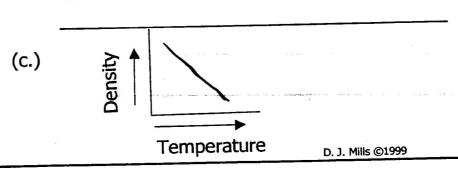
2. Density in Relation to Temperature

Example 2 = particle of matter/1 gram



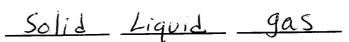
- (1.) volume <u>increases (expands)</u>
- (2.) mass <u>remains the same</u>
 (3.) density <u>decreases</u>
- (b.) Relationship: As temperature.

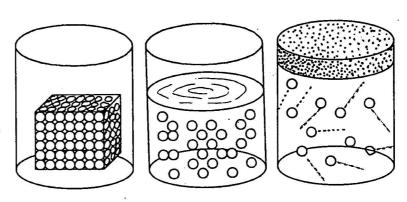
increases, density decreases



3. Density in Relation to State of Matter

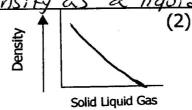
Model Problem (a.)

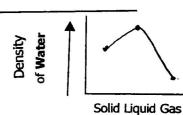




- (b.)
- (c.) Water

(d.) (1)

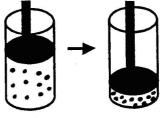




4. Density in Relation to Pressure (on a gas)

(a.) Model Problem

Key = particle of matter



- (1.) volume -
- decreases
- (2.) mass -
- remains
- (3.) density -
- increases

pressure increases (b.) Relationship: As

density increases (c.) Pressure D. J. Mills @1999

I. Temperature - Average Kinetic energy of a material

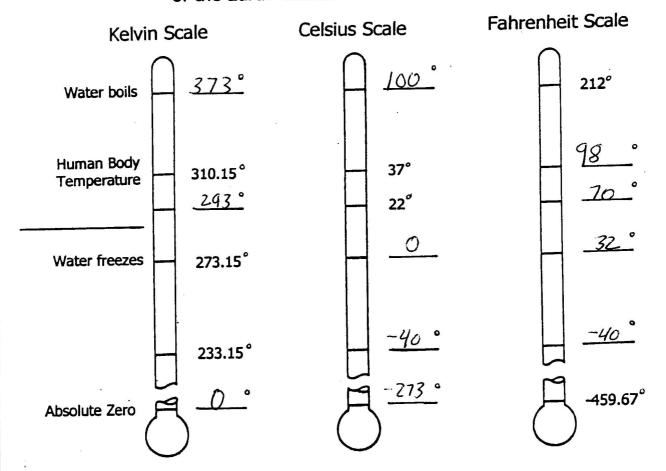
1. Instrument - Thermometer

(a.) Fahrenheit 2. Units – degrees:

(b.) <u>Celsius</u>

(c.) Kelvin

of the Earth Science Reference Tables.



4. Temperature Conversion Formulas:

$$^{\circ}K = ^{\circ}C + 273.15$$

$$^{\circ}C = \frac{^{\circ}F - 32}{1.8}$$

$$^{\circ}F = 1.8 (^{\circ}C) + 32$$